Nonpeptidic $\alpha_*\beta_3$ Integrin Antagonist Libraries: On-Bead Screening and Mass Spectrometric Identification without Tagging**

Christoph Gibson, Gábor A. G. Sulyok, Diane Hahn, Simon L. Goodman, Günter Hölzemann, and Horst Kessler*

Over the past decade, combinatorial synthesis has progressed from being a little-known fringe area to an important technique in drug research.^[1] The split synthesis in particular enables comprehensive substance libraries to be produced rapidly.^[2] Such libraries also enable those resin particles with attached active compounds to be characterized by biological on-bead evaluation.[3] However, the coding technique required for the identification of a particular selected substance usually reduces the efficiency of this approach.[1] Herein, we describe the combinatorial solid-phase synthesis of uncoded diacylhydrazine libraries and their biological on-bead evaluation using a soluble $\alpha_{\nu}\beta_{3}$ integrin receptor. Orthogonal anchoring with a photolabile linker made it possible to photolytically cleave the compounds from selected beads and to identify them reliably using mass-spectrometric analysis (MSⁿ) on the basis of fragmentation patterns. The degree of affinity of the selected diacylhydrazines to the $\alpha_{v}\beta_{3}$ receptor was confirmed through receptor-binding studies carried out on the isolated compounds, which underlines the success of our strategy. To our knowledge, this is the first successful application of the "one bead-one compound" concept as applied to uncoded, nonpeptidic compound libraries.[3]

The inhibition of the $\alpha_{\nu}\beta_{3}$ integrin receptor is regarded as a promising goal in the therapy of various pathophysiological processes such as tumor-induced angiogenesis, restenosis, osteoporosis, and acute renal failure. The amino acid sequence Arg-Gly-Asp (RGD), which can be recognized by the $\alpha_{\nu}\beta_{3}$ receptor and by at least ten additional integrins, has been used as a lead structure in the development of $\alpha_{\nu}\beta_{3}$ inhibitors.

In previous work, we have been able to show that the substitution of the glycine in cyclic $\alpha_v \beta_3$ -selective RGD peptides can take place with an aza-glycine whilst maintaining both affinity and selectivity. [6] Based on these results, we developed a diacylhydrazine library that meets all require-

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Supporting information for this article is available on the WWW under http://www.angewandte.com or from the author. ments of the "one bead–one compound" concept. [3] To achieve this, the lead structure Arg-Gly-Asp was transformed into a RGD mimetic, which could be assembled step-by-step on a solid support according to the Fmoc strategy [7] with the building blocks $\mathbf{A} - \mathbf{D}$ (see Figure 1). The RGD mimetics were anchored to the resin with a photolinker [8] to achieve maximum orthogonality to the various reaction conditions of the solid-phase synthesis.

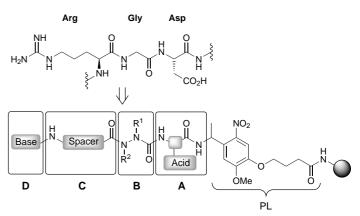


Figure 1. Transformation of the RGD sequence into a modularly assembled RGD mimetic.

The building blocks used for the synthesis of the library are shown in Figure 2.[9] The synthesis of activated aza-substituted building blocks B^{1-3} and their incorporation onto the solid support has recently been described by us.[10] The building blocks of groups A and C were selected in such a way that each molecular mass occurred only once within one group: This enabled the structure assignment of a selected RGD mimetic using MS^n (see below). The sequence for the synthesis of the RGD mimetics by applying the split method^[2] is shown in Scheme 1. Firstly, the building blocks A^{1-5} were attached to TentaGel Macrobeads under standard conditions. Building block A^6 could be synthesized on the solid support using a submonomeric approach. To achieve this, TentaGel Macrobeads were acylated with 3-(chloromethyl)benzoyl chloride and the resulting resin-bound benzyl chloride was treated with tert-butyl-3-aminopropionate to produce the resin-bound building block A^6 . After removal of the Fmoc protecting group, the activated aza-building blocks B^{1-3} were introduced.[10] Basic deprotection, coupling of the spacer building blocks C^{1-10} , and subsequent removal of the temporary and permanent protecting groups resulted in ten amino-RGD mimetic libraries, each comprising 33 compounds.[11] Through guanylation with D^1 or pyrimidylation with D^2 and subsequent deprotection, two additional libraries each comprising 330 compounds were obtained.[12]

For the quality control of the RGD mimetic libraries and after the biological on-bead evaluation, the compounds were photolytically cleaved from individual beads and characterized subsequently with LC-MS and/or ESI-MSⁿ. In order to assess the suitability of this strategy, several isolated synthesized aza-RGD mimetics were investigated by MSⁿ. Figure 3 illustrates the results of the MS² analysis of the aza-RGD mimetics 1-3. The MS² spectra of the $[M+H]^+$ ions (Figure 3,

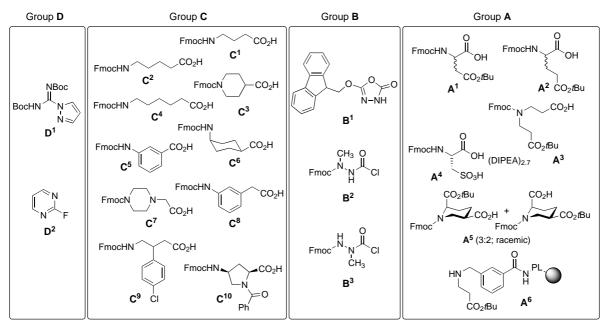
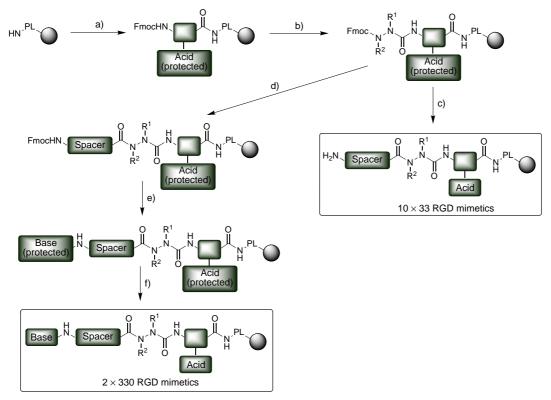


Figure 2. Building blocks used for the synthesis of the RGD mimetic libraries.



Scheme 1. a) ${\bf A}^{1-5}$ (3.0 equiv), HATU (2.8 equiv), collidine (30 equiv), DMF, RT, 3 h; ${\bf A}^6$ 3-(chloromethyl)benzoyl chloride (6.8 equiv), DIEA (14 equiv), CH₂Cl₂, RT, 75 min; tert-butyl-3-aminopropionate (150 equiv), DMF, 45 °C, 12 h; b) 20 % piperidine in DMF; ${\bf B}^1$ (3.1 equiv), CH₂Cl₂, RT, 3 h; ${\bf B}^2$ (5.1 equiv), DIEA (6.0 equiv), CH₂Cl₂, RT, 2.5 h; ${\bf B}^3$ (5.1 equiv), DIEA (5.5 equiv), DMF, RT, 15 h; c) 20 % piperidine in DMF; ${\bf C}^{1-10}$ (3.0 equiv), HATU (2.8 equiv), collidine (30 equiv), DMF, RT, 12 h; 20 % piperidine in DMF; ${\bf C}^{1-10}$ (3.0 equiv), HATU (2.8 equiv), collidine (30 equiv), DMF, RT, 12 h; e) 20 % piperidine in DMF; ${\bf D}^1$ (19 equiv), CHCl₃, 50 °C, 20 h or ${\bf D}^2$ (15 equiv), DIEA (15 equiv), DMF, RT, 1 d, then ${\bf D}^2$ (30 equiv), 5 % BF₃·Et₂O in DMF, RT, 7 d; f) 50:50:5 CH₂Cl₂:TFA:TIPS, RT, 1.5 h; 20 % DIEA in CH₂Cl₂. Fmoc = fluoren-9-ylmethoxycarbonyl, DMF = N,N-dimethylformamide, HATU = O-(7-azabenzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate, DIEA = diisopropylethylamine, TIPS = triisopropylsilane.

left) and the $[M+Na]^+$ ions (Figure 3, right) show clearly defined signals of the fragment ions of B_B and/or A_B types. This information enables building block **A** to be identified. In

particular, the MS² spectra of the $[M+Na]^+$ ions of isomeric compounds **2** and **3** show clearly different intensities of the B_B and A_B fragment ions: Compound **2** containing aza-Sar

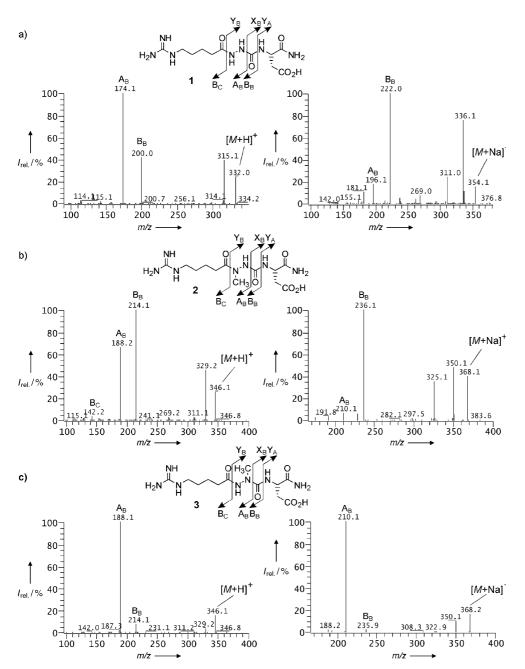


Figure 3. MS^2 analysis of aza-RGD mimetics containing aza-Gly (a), aza-Sar (b), and aza-Ala (c). Note that the $[M+Na]^+$ ion of the aza-Sar-containing compound **2** predominantly yields the fragment ion of the type B_B , whereas the $[M+Na]^+$ ion of the aza-Ala-containing compound **3** predominantly yields the fragment ion of the type A_B . In order to define the fragment ions unequivocally, they were subscripted with the building block, which is adjacent to the fragmentation position in the fragment ion. RA = relative abundance.

decomposed with 10:1 selectivity to the B_B fragment ion and compound 3 containing aza-Ala decomposed with approximately the same selectivity to the A_B fragment ion. Thus, one can differentiate clearly between regioisomeric building blocks **B** aza-Sar (B^2) and aza-Ala (B^3). In addition, the weak signal of the B_C fragment ion can be recognized in the MS^2 spectrum of the $[M+H]^+$ ion of 2. On the other hand, this fragment ion appears in all the MS^3 spectra of the B_B and A_B fragment ions of protonated species 1-3 in noticeably higher intensities and, hence, enables building blocks **B** and **C** to be assigned. Consequently, our diacyl hydrazines

may be fully sequenced using mass spectrometry and unequivocally identified. The reliability of this method was confirmed by applying MSⁿ analysis to additional aza-RGD mimetics. In this context, the fragment ions of the types X and Y, that correspond to the fragment ions of the types A and B, were also observed.

The ionization properties of the amino- as well as the guanyl-RGD mimetics were considerably improved by introduction of an "N-terminal" Boc protective group. Most of the aza-RGD mimetics investigated exhibited a high degree of purity; about one third of the LC-MS-spectra additionally contained a compound lighter by m/z 18 with a relative proportion of 30-70%.

For biological on-bead evaluation, the resin-bound RGD mimetics were first incubated with the soluble, biotinylated $\alpha_{\rm v}\beta_{\rm 3}$ receptor and subsequently with a monoclonal antibiotin alkaline phosphatase conjugate. If a compound was recognized by the receptor, the respective bead could be stained with the well known alkyl phosphatase substrate 5-bromo-4-chloro-3-indolyl phosphate disodium (BCIP).[13] No positive beads were found in the amino-libraries. Figure 4 depicts the beads of the guanylated and pyrimidylated libraries after on-bead evaluation with the $\alpha_{\nu}\beta_{3}$ receptor. In the guanylated library, there were four intensely stained and ten weakly stained

beads; in the pyrimidylated library, two weakly stained beads were observed.

Using LC-MS and MSⁿ, all RGD mimetics bound to the beads and designated positive could be characterized unequivocally. The four intensely stained beads all carried aromatic compound 5 (Table 1). Of the ten weakly stained beads from the same library, compound 5 was detected once, compound 6 five times, compound 7 three times, and compound 8 once. Compound 9—analogous to RGD mimetic 5—was detected on the two beads selected from the pyrimidylated library. For the determination of the inhibition

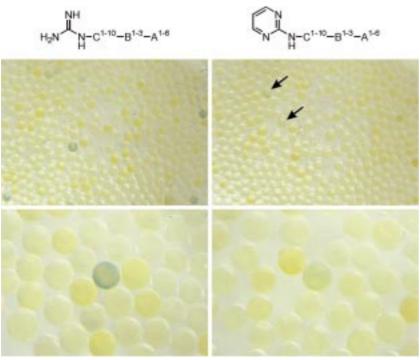


Figure 4. TentaGel Macrobeads after the on-bead receptor assay with the $\alpha \beta_3$ integrin. Above: fivefold enlargement, below: fifteenfold enlargement.

constants with respect to the isolated $\alpha_{\nu}\beta_{3}$, $\alpha_{\rm v}\beta_{\rm 5}$, [4b] and $\alpha_{\rm IIb}\beta_{\rm 3}$ [14] receptors, the selected RGD mimetics were synthesized on Rink amide 4-methyl-benzhydrylamine (MBHA) resin as L- and D-enantiomers, respectively. All RGD mimetics showed a measurable activity on the $\alpha_{\rm v}\beta_3$ receptor but the Lenantiomers, however, exhibited throughout a higher affinity to the $\alpha_{\nu}\beta_{3}$ receptor than the respective D-enantiomers. With an IC₅₀ value of 150 nm, compound 5 showed, in contrast to the other selected aza-RGD mimetics, the highest affinity to the $\alpha_{\nu}\beta_{3}$ receptor; thus, the color intensity of the positive beads clearly correlates with the affinity of the resin-bound RGD mimetics to the receptor.

Only the L-isomers of **5** and **6** showed a weak affinity to the $\alpha_{\rm v}\beta_5$ receptor, both being less pronounced when compared to the $\alpha_{\rm v}\beta_3$ receptor. All selected aza-RGD mimetics inhibited the binding of the platelet receptor $\alpha_{\rm IIb}\beta_3$ to fibrinogen with IC₅₀ values of >100 μ m. Thus, all the selected

Table 1. Inhibition behavior of the different RGD mimetics with regard to binding of vitronectin to the isolated $\alpha_{\nu}\beta_{3}$ or $\alpha_{\nu}\beta_{5}$ receptor, as well as the binding of fibrinogen to the isolated $\alpha_{\Pi b}\beta_{3}$ receptor. The linear peptide GRGDSPK 4 was chosen as standard.

No.	Structure and frequency	Isomer of A		IC ₅₀ [μM]	
			$a_{ m v}eta_3$	$a_{ m v}eta_{ m 5}$	$\alpha_{ ext{IIb}}eta_3$
4	GRGDSPK		0.40	42	1.1
5	H_2N N N N N N N N N N	L-Asp-NH ₂ D-Asp-NH ₂	0.15 7.2	7.2 > 100	> 100 > 100
6	H_2N N N N N N N N N N	L-Asp-NH ₂ D-Asp-NH ₂	3.1 50	57 > 100	> 100 > 100
7	H_2N H_2N H_3N H_4 H_5 H_5 H_5 H_5 H_5 H_5 H_6 H_7 H	L-Asp-NH ₂ D-Asp-NH ₂	5.0 41	> 100 > 100	> 100 > 100
8	H_2N H_2 N H_2 N	$\begin{array}{c} \text{L-Asp-NH}_2 \\ \text{D-Asp-NH}_2 \end{array}$	6.8 47	> 100 > 100	> 100 > 100
9	$ \begin{array}{c c} N & H & O \\ N & H & O \\ N & NH_2 & 2 \text{ of } 2 \end{array} $	L-Asp-NH ₂ D-Asp-NH ₂	15 53	> 100 > 100	> 100 > 100
10	H_2N N N N N N N N N N		0.0026	0.280	8.3

aza-RGD mimetics showed a clear preference for the $\alpha_{\rm v}\beta_{\rm 3}$ receptor.

Lead structure **5** was found to exhibit a relatively high degree of polarity; an unfavorable pharmacokinetic profile can thus be expected for this compound. However, it has been shown in the past that a hydrophobic residue in the β -position to the carboxy group is tolerated by the $\alpha_v \beta_3$ receptor. [4d] For this reason, we replaced the terminal carboxamide group with a phenyl residue. The less polar RGD mimetic **10** exhibited similar selectivity and considerably increased activity on the $\alpha_v \beta_3$ receptor compared to the polar mimetic **5**.

The results obtained would appear to support our concept for identifying new, low molecular weight integrin ligands through application of combinatorial solid-phase synthesis, biological on-bead evaluation, and mass spectrometry to the selected compounds.

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- a) F. Balkenhohl, C. von dem Bussche-Hünnefeld, A. Lansky, C. Zechel, Angew. Chem. 1996, 108, 2436–2488; Angew. Chem. Int. Ed. Engl. 1996 35, 2289–2337, and references therein; b) P. Wentworth, K. D. Janda, Curr. Opin. Biotechnol. 1998, 9, 109–115, and references therein; c) R. S. Houghten, C. Pinilla, J. R. Appel, S. E. Blondelle, C. T. Dooley, J. Eichler, A. Nefzi, J. M. Ostresh, J. Med. Chem. 1999, 42, 3743–3778, and references therein.
- [2] A. Furka, F. Sebestyen, M. Asgedom, G. Dibo, Int. J. Pept. Protein Res. 1991, 37, 487 – 493.
- [3] K. S. Lam, S. E. Salmon, E. M. Hersh, V. J. Hruby, W. M. Kazmierski, R. J. Knapp, *Nature* 1991, 354, 82–84.
- [4] a) R. Haubner, D. Finsinger, H. Kessler, Angew. Chem. 1997, 109, 1440–1456; Angew. Chem. Int. Ed. Engl. 1997, 36, 1374–1389, and references therein; b) J. Samanen, Z. Jonak, D. Rieman, T.-L. Yue, Curr. Pharm. Des. 1997, 3, 545–584, and references therein; c) M. S. Goligorsky, H. Kessler, V. I. Romanov, Nephrol. Dial. Transplant. 1998, 13, 254–263, and references therein; d) R. M. Scarborough, Curr. Med. Chem. 1999, 6, 971–981, and references therein.
- [5] a) E. Ruoslahti, M. D. Pierschbacher, *Cell* 1986, 44, 517-518; b) E.
 Ruoslahti, M. D. Pierschbacher, *Science* 1987, 238, 491-497; c) T. A.
 Springer, *Nature* 1990, 346, 425-434; d) M. Pfaff in *Integrin Ligand Interaction* (Ed.: J. A. Eble), Springer, Heidelberg, 1997, pp. 101-121.
- [6] a) J. Wermuth, PhD Thesis, Technische Universität München (Germany), 1996; b) J. S. Schmitt, PhD Thesis, Technische Universität München (Germany), 1998.
- [7] G. B. Fields, R. L. Noble, Int. J. Pept. Protein Res. 1990, 35, 161 214, and references therein.
- [8] a) C. P. Holmes, J. Org. Chem. 1997, 62, 2370 2380; b) C. P. Holmes,
 D. G. Jones, J. Org. Chem. 1995, 60, 2318 2319.
- [9] For the synthesis of building blocks A³⁻⁶, see the Supporting Information. Although the regioisomers of compound A⁵ could be separated by HPLC, they were used as a mixture of isomers for the library synthesis.
- [10] C. Gibson, S. L. Goodman, D. Hahn, G. Hölzemann, H. Kessler, J. Org. Chem. 1999, 64, 7388 – 7394.
- [11] Six different building blocks in group A were used for the library synthesis, whereas building blocks A¹ and A² were used as a racemate and A⁵ as a racemate and a mixture of regioisomers. In consideration of the isomers in group A, 11 different building blocks have been employed.
- [12] The guanylation was performed as per: Y. Wu, G. R. Matsueda, M. Bernatowicz, *Synth. Commun.* 1993, 23, 2055 3060; the pyrimidylation was performed as per: C. Gibson, H. Kessler, *Tetrahedron Lett.* 2000, 41, 1825 1728.
- [13] The substrate BCIP is usually used together with the oxidation agent 4-nitro-blue-tetrazolium (NBT). However, NBT proved to be incompatible with our libraries because it reacted with some of the resinbound compounds and led to false-positive results.
- [14] I. Ojima, S. Chakravarty, Q. Dong, Bioorg. Med. Chem. 1995, 3, 337 360, and references therein.

The First Synthetic Application of a Monooxygenase Employing Indirect Electrochemical NADH Regeneration**

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In memory of Eberhard Steckhan

One of the most important challenges in applying monooxygenase reactions in vitro is to find an effective regeneration system for the necessary co-enzyme (mostly NAD(P)H). The well-established methods for the regeneration of the nicotinamide co-enzyme mainly consist of an enzyme-coupled approach utilizing formate dehydrogenase^[1, 4c] (for NAD(P)H) or glucose-6-phosphate dehydrogenase (for NADPH).^[2] Additionally, non-enzymatic redox catalysts have been developed and successfully applied to NAD(P)Hdependent dehydrogenases.^[3] Thus only the producing enzyme and a mediator together with the electrode, as a source of reducing equivalents, are needed.

Here we report on the first application of an isolated monooxygenase with an indirect electrochemical regeneration of NADH. The enzyme employed is the 2-hydroxybiphenyl-3-monooxygenase (HbpA, E.C. 1.14.13.44), a member the class of flavine-dependent monooxygenases, from *P. azelaica*. The homotetramer with a total mass of 256 kDa catalyzes the specific *ortho*-hydroxylation of several α -substituted phenol derivatives (Scheme 1). To the best of our knowledge no chemical counterpart with comparable specificity is known.

Scheme 1. Specific *ortho*-hydroxylation of α -substituted phenols catalyzed by 2-hydroxybiphenyl-3-monooxygenase. R = alkyl (Et, Pr, iPr), aryl (Ph, 2-HOC₆H₄), Hal (F, Cl, Br).

For the regeneration of NADH we applied the [Cp*Rh(bpy)Cl]Cl complex which had been developed in our group $(Cp*=C_5Me_5; bpy=2,2'-bipyridine)$. The corresponding hydridorhodium complexes, which can be generated either electrochemically by cathodic reduction at -750 mV (versus $Ag/AgCl_{sat}$) or chemically with formate, transform $NAD(P)^+$ efficiently into the enzymatically active 1,4-NAD(P)H form $^{[3,5]}$ (Scheme 2). The conversion rates

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